

**Tentamination      FTV 2**

***Friday August 25, 2006***

**09:00–12:00 / 5118.-152**

**Mention on the first page:**

- Name
- Address
- Student number
- Day of birth
- Study
- Year of registration RuG

**And on the following pages:**

- Your Name

Question 1:

- a) Derive from the expression for the penetration depth that instationary heat transfer at short times can be described by the expression:

$$Nu = 0,57 Fo^{\frac{1}{2}}$$

- b) For longer times the Nusselt number becomes constant. Derive for this situation an expressing for the average temperature of a body as function of time.

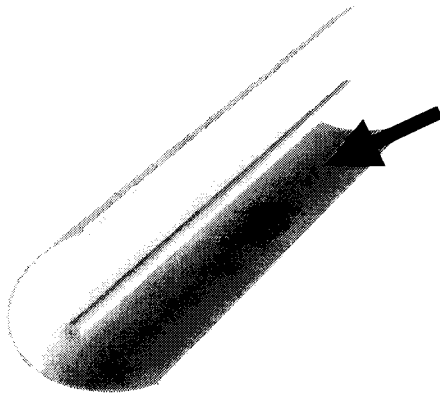
Question 2:

A tube with a diameter of 1,25 cm is placed at the focus of a solar collector. The solar energy received by the tube is  $\Phi''_s = 24 \text{ kW/m}^2$ . Water is heated while it flows through the tube with a velocity of 1 m/s. Ambient temperature and the water temperature at the inlet are both  $25^\circ\text{C}$ . The water leaves the solar collector with a temperature of  $100^\circ\text{C}$ .

The energy losses of the tube are due to radiation to the surroundings (at  $25^\circ\text{C}$ ) and free convection of the ambient air.

- Derive a differential equation for the axial temperature profile in the tube (including all effects).
- Make an estimation of the relative importance of the heat losses due to radiation and free convection.
- Determine the length of the tube (and collector) needed for water to reach an exit temperature of  $100^\circ\text{C}$ .
- How much extra length do you need to obtain steam of  $100^\circ\text{C}$ .

The heat transfer coefficient due to free convection equals  $25 \text{ W/m}^2 \text{ K}$ . Find all other relevant quantities yourself and state clearly the values you use.



Question 3:

At a recycle plant an oil is collected that has been polluted with a chlorinated carbohydrate. One wants to remove the pollutant from the oil by extraction with water. The solubility of the pollutant in water is 200 fold its solubility in oil ( $m=200$ ). For the extraction the oil is dispersed as droplets in the water. The mass transfer inside and outside the droplets occurs only by diffusion.

- a. Show that the resistance to mass transfer for long times lies mainly in the oil phase. (Hint: which Sherwood relations can be used outside and inside the droplet?)
- b. Show that the same can be concluded for short times. Neglect the curvature of the surface of the droplet.
- c. Calculate the time needed to decrease the concentration of the pollutant in the oil phase with a factor of 100. Assume therefore that the concentration in the water phase remains negligible.

Given:

Diffusion coefficient of the pollution in oil:  $D_0=2 \cdot 10^{-10} \text{ m}^2/\text{s}$

Diffusion coefficient of the pollution in water:  $D_w=6 \cdot 10^{-10} \text{ m}^2/\text{s}$

Diameter of the oil droplets:  $d = 2 \cdot 10^{-4} \text{ m}$